Effect of Tooth-whitening Procedures on Stained Nanohybrid Composites: Spectrophotometric Analysis

Sruthi M Sunildath, Josey Mathew, Liza George, Sinju Paul, Dhanya John, Aleesha Joy

ABSTRACT

Aim: To evaluate and compare the staining of a nanohybrid composite resin by coffee and turmeric powder and to determine the degree of lightening that could be achieved by using contemporary self-applied at-home products and in-office whitening techniques.

Materials and methods: Thirty-six samples of standardized dimensions (10 mm diameter, 2 mm thick disks) were fabricated using GC solare sculpt A2 shade. The color of each specimen was measured (L, a, b values) at baseline with SpectroShade Micro. Samples were randomly allocated into three groups of coffee, turmeric solution, distilled water, and placed in 50 mL of each test solution. The color of each specimen was measured (L, a, b values) after 7 days to determine the color change. Each group was further divided into four subgroups according to the whitening treatment. Subgroup A—crest 3D white toothpaste (applied twice daily for 7 days), subgroup B—Polanight (45 minutes a day for 4 days), subgroup C—polaoffice (8 minutes each for three times), and subgroup D—Shofu polishing kit (applied once). After placing all specimens in distilled water for 24 hours, the color was measured again using a spectrophotometer. The effectiveness of whitening protocols was assessed by subtracting baseline color index values from those measured after whitening.

Results: Group II (turmeric solution) showed greater color change (ΔE) after staining for 1 week. There was significant difference in color changes imparted following whitening treatments with crest 3D white toothpaste after staining with coffee and turmeric solution.

Conclusion: Within the limitations of this in vitro study, it was observed that the whitening treatment with crest 3D white toothpaste is an effective method for reversing the color changes imparted by coffee and turmeric on the nanohybrid composite resin.

Clinical significance: A composite resin stained by coffee and turmeric can be lightened to a clinically perceptible level by crest 3D white toothpaste.

Keywords: Laboratory research, Nanohybrid composite, Peroxide, Sodium hexametaphosphate, Spectrophotometer, Tooth whitening.


Source of support: Nil

Conflict of interest: None

INTRODUCTION

A beautiful smile can impact the level of self-esteem and it makes the person more confident. Teeth that are intact, properly aligned, and not discolored have a positive psychological impact. There are many factors in determining the esthetics of a smile of which tooth color is an essential component in it.

Resin-based restoratives are increasingly being used in dentistry, mainly because of their aesthetic quality and good physical properties. Composites can mimic the natural color and shade of the tooth. The nanohybrid composite combines nanomeric and conventional fillers and they offer high translucency, high polish, superior gloss, and adequate mechanical properties. These superior esthetic properties of the nanohybrid composite resin enable them to be used as anterior restorations.

Exogenous colorants from food and drinks can stain composites due to adsorption. Over many years, studies have been conducted to measure these stainings by beverages such as fruit juices, tea, coffee, soy sauce, and cola and food spices. Coffee and turmeric that have various medicinal benefits are commonly used by Indians. Both materials have the propensity to stain esthetic restorative materials.

Evaluation of staining of composites can be done by visual and instrumental techniques. Visual techniques can cause errors due to subjective variations in color perception. A spectrophotometer and a colorimeter can be used as alternatives as they are less subjected to variations.

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Aim

The purpose of this study is to evaluate and compare

• The staining of a nanohybrid composite resin (GC solare sculpt, Japan) by coffee and turmeric powder.
• The degree of lightening that could be achieved by using contemporary self-applied at-home products (SDI Polanight and crest 3D white toothpaste) and in-office techniques (SDI polaoffice and Shofu polishing kit).
**Materials and Methods**

Samples of standardized dimensions (10 mm diameter, 2 mm thick disks) were fabricated using GC solare sculpt A2 shade \((n = 36)\) (Fig. 1). The color of each specimen was measured \((L, a, b)\) values at baseline with SpectroShade Micro II (MHT Switzerland).

**Staining**

Samples were randomly allocated into three groups of coffee, turmeric solution, and distilled water and placed in 50 mL of each test solution. Coffee solution was prepared by using 10 g of coffee powder to 50 mL of hot water. Turmeric solution was prepared by adding 1.0 g of turmeric powder to 50 mL of boiling distilled water. All solutions were freshly prepared and replaced daily for 7 days. The color of each specimen was measured \((L, a, b)\) values after 7 days to determine the color change (Fig. 2).

Total \(\Delta E^*\) color change was calculated according to the expression:

\[
\Delta E^* = \left[ (\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2 \right]^{1/2}
\]

where \(L^*\) is lightness \((-L^* = \text{black}; +L^* = \text{white})\), \(a^*\) is green-red \((-a^* = \text{green}; +a^* = \text{red})\), \(b^*\) is blue-yellow \((-b^* = \text{blue}; +b^* = \text{yellow})\).

**Whitening Treatment**

Each group was further divided into four subgroups according to the whitening treatment: subgroup A—crest 3D white toothpaste—Procter and Gamble Cincinnati, OH, United States (applied twice daily for 7 days—Fig. 3); subgroup B—Polanight 22%—SDI Polanight Australia (45 minutes a day for 4 days—Fig. 4); subgroup C—Polaoffice 35% hydrogen peroxide, Polaoffice Bleaching, SDI Limited, Bayswater, Victoria, Australia (8 minutes each for three times—Fig. 5); and subgroup D—Shofu polishing kit (polished with a polishing disk on a slow-speed handpiece for 10 seconds at moderate pressure—Fig. 3). Each of the whitening products were applied in a manner simulating at-home or in-office treatment and as per manufacturer's instructions. After 24 hours in distilled water, all specimens were measured for color change using the spectrophotometer and the effectiveness of whitening protocols was assessed by subtracting baseline color index values from those measured after whitening.

**Statistical Analysis**

Data were analyzed using the SPSS version 24 software and a confidence level of 95% was set \((p < 0.05)\). The one-way ANOVA was used for comparison among the groups. The Tukey HSD test was done to compare among subgroups.

**Results**

Spectrophotometric analysis of composite specimens after staining for 1 week showed greater \(\Delta E\) value for group II, followed by group I (Table 1).

The Tukey HSD test was done to compare two staining methods and it was observed that there was statistically significant difference between control and staining groups. But there was no significant difference between group I and group II (Table 2).

The one-way ANOVA was used to compare among groups after whitening treatments and in all the groups the \(p\) value was less than 0.05. There was significant difference in color changes imparted by whitening treatments after staining by three methods (Table 3).

Multiple comparison between subgroups was done using the Tukey HSD test. Pairwise comparisons were significant between all subgroups except between subgroups B and D (polanight and polish) in the coffee staining group. In the turmeric group, subgroup A (crest 3D white toothpaste) showed statistically significant difference with all the other groups \((p < 0.05)\) (Table 4).
Composites are constantly exposed to saliva, beverages, and food stains, so it is important to determine its staining susceptibility. Discoloration of composite has been reported to occur at three levels: surface (or external), subsurface, and body (or intrinsic). Subsurface stains are attributed to surface stains diffusing into superficial resin layers, with pigments attaching to potentially reactive chemicals within the composite. Intrinsic stain mostly is ascribed to physicochemical reactions occurring deep within the composite.\textsuperscript{10}

Resin matrix chemistry, filler loading, particle size, and degree of conversion are known to affect water sorption of the composite, therefore the nature and severity of composite staining are different for each manufacturer’s product.\textsuperscript{11} Nanohybrid composites are those that contain milled glass fillers and discrete nanoparticles (40–50 nm) as the dispersed phase. They provide better mechanical strength and surface finish than hybrid composites.\textsuperscript{12}
Staining of resins by fluid pigments and beverages is caused by adsorption or absorption (the uptake of substances into or through tissues) of colorants by resins.\textsuperscript{13} Conjugated diarylhepnoiids like curcumin are responsible for orange color and staining of turmeric solution.\textsuperscript{12} Staining by coffee is due to tannic acid and discoloration from coffee is due to both the surface adsorption and absorption of colorants.\textsuperscript{14}

For standardized and reproducible evaluation of color changes of restorative materials, spectrophotometers are used for analysis of \( L^*a^*b^* \) values according to the CIELAB system.\textsuperscript{15} It contains monochromators and photodiodes that measure the reflectance curve every 10 nm or less.\textsuperscript{16} The Commission Internationale del’Eclairage (CIELAB) color coordinates system provides information about location of object color in a uniform three-dimensional color space. It quantifies the color in terms of three coordinate values, \( L^*, a^*, \) and \( b^* \). Here \( L^* \) represents brightness or lightness (value) and \( a^* \) and \( b^* \) serve as numeric correlates both for hue and chroma.\textsuperscript{17} The \( a^* \) and \( b^* \) values represent position on a red/green and yellow/blue axis, respectively: \(+a^* = \text{red}, -a^* = \text{green}, +b^* = \text{yellow}, \text{and} -b^* = \text{blue} \). Values of \( \Delta E \) between 0 and 2 are imperceptible, those between 2 and 3 are just perceptible, values from 3 to 8 are moderately perceptible, and values above 8 are markedly perceptible.\textsuperscript{18}

A \( \Delta E \) value of 3.7 or less is considered to be clinically acceptable.\textsuperscript{19,20} Stober et al. concluded that turmeric solution and red wine caused the most severe discoloration (\( \Delta E > 10 \)) than tea, coffee, and mouth rinse over 4 and 8 weeks.\textsuperscript{21} The study by Karadars showed clinically perceptible color change in composite when immersed in black tea or coffee.\textsuperscript{22} In the present study, color change in the dental resin material was observed with turmeric and coffee (\( \Delta E > 8 \)); greater staining was observed with turmeric (\( \Delta E = 9.94 \)). But there is no statistically significant difference between groups.

Polaooffice contains 35% hydrogen peroxide, and Polanight contains 22% carbamide peroxide. When activated by heat, light, or pH, hydrogen peroxide (\( \text{H}_2\text{O}_2 \)) dissociates via cleavage of O–H and O–O bonds to produce \( \text{H}^* \), \( \text{OOH}^* \), and \( 2\text{OH}^* \) radicals that are capable of decomposing conjugated double bonds and ring structures present within stain.\textsuperscript{23} Hydrogen peroxide and urea are the breakdown products of carbamide peroxide, and urea further dissociates into carbon dioxide and ammonia. This creates an alkaline environment capable of producing the perhydroxyl anion (\( \text{HO}_2^* \)), a compound shown to further the whitening effect.\textsuperscript{24} In this study, Polaooffice showed significant whitening compared to other whitening treatments in coffee group.

The crest 3D white toothpaste showed significant difference with all other whitening treatments in both coffee and turmeric groups. Peroxide-based systems were not able to whiten the composite as the whitening toothpaste. The crest 3D white toothpaste contains sodium hexametaphosphate that has multiple binding sites translates to greater coverage of the tooth surface, increasing its potential retention and substantivity.\textsuperscript{25,26} Numerous clinical and \textit{in vitro} studies have demonstrated sodium hexametaphosphate’s stain-control benefits in dentifrice and chewing gum delivery systems.\textsuperscript{27–31} Study by Manis et al. showed that whitening toothpaste influenced the color stability of the composite resin.\textsuperscript{32} Abrasive action during brushing and presence of sodium hexametaphosphate might have contributed to the better results by the crest 3D white toothpaste.

<table>
<thead>
<tr>
<th>Groups</th>
<th>Subgroups</th>
<th>n</th>
<th>Mean</th>
<th>Std deviation</th>
<th>( F )</th>
<th>Sig.</th>
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<tr>
<td></td>
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<td></td>
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</table>

Table 3: Mean and standard deviation of \( \Delta E \) after whitening treatment of the stained composite

Table 4: Multiple comparison among subgroups using the Tukey HSD test

<table>
<thead>
<tr>
<th>Subgroups (I)</th>
<th>Subgroups (J)</th>
<th>Mean difference (I – J)</th>
<th>Sig.</th>
</tr>
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<td>0.000</td>
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<tr>
<td></td>
<td>C</td>
<td>−1.19667*</td>
<td>0.000</td>
</tr>
<tr>
<td></td>
<td>D</td>
<td>−1.23000*</td>
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<tr>
<td>B</td>
<td>C</td>
<td>1.19667*</td>
<td>0.000</td>
</tr>
<tr>
<td></td>
<td>D</td>
<td>−1.25333*</td>
<td>0.000</td>
</tr>
<tr>
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<td></td>
<td>−0.03333</td>
<td>0.995</td>
</tr>
<tr>
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<td>D</td>
<td>2.45000*</td>
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</tr>
<tr>
<td></td>
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<td>1.25333*</td>
<td>0.000</td>
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<td></td>
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<td>1.22000*</td>
<td>0.000</td>
</tr>
<tr>
<td>D</td>
<td></td>
<td>−1.22000*</td>
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</tr>
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<td></td>
<td>0.03333</td>
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Whitening Procedures on Stained Nanohybrid Composites: Spectrophotometer

CONCLUSION

Within the limitations of this in vitro study, it was observed that the whitening treatment with the crest 3D white toothpaste is an effective method for reversing the color changes imparted by coffee and turmeric on the nanohybrid composite resin.

REFERENCES